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Synthetic Investigations in Diterpenoids

For some time we have been carrying out experiments to develop stereospecific syntheses for the diterpenoids. In view of a recent communication by Stork and Schwlenberg¹ on the synthesis of dl-dehydroabietic acid, we wish to report the results of our own work on the synthesis of DL-6-methoxypodocarpane (VI).

1-Methyl-7-methoxy-3, 4-dihydronaphthalene² (I) was oxidized in acetic acid solution with red lead oxide³ and the crude diacetate rearranged in dilute alcoholic sulphuric acid to 1-methyl-7-methoxytetralone-2 (II), b.p. 125–126°/0·8 mm, n_D^{27} 1·5730 (yield 72%) (calculated for C_{12} H_{14} O_2 : C, 75·8; H, 7·4; found: C, 76·1; 7·4), semicarbazone m.p. 191–192° (calculated for $C_{13}H_{17}O_2N_3$: N, 17·0; found: N, 17·2). Condensation of the β -tetralone (II) with 4-diethylaminobutanone-2-methiodide⁴ afforded 6-methoxy-2-keto-4a-methyl-2, 3, 4, 4a, 9, 10-hexahydrophenanthrene (III), b.p. 115–120° (air-bath)/0·004 mm as a very viscous oil (yield 62%) (calculated for $C_{16}H_{18}O_2$: C, 79·3; H, 7·5; found: C, 78·84; H, 7·5), 2, 4-dinitrophenylhydrazone, m.p. 241–242° (dec.) (calculated for $C_{13}H_{17}O_2N_3$: C, 62·6; H, 5·3; found: C, 62·9; H, 5·5). This (III) was then methylated with methyl iodide

in the presence of potassium tert-butoxide⁵ to give 1,1dimethyl-2-keto-6-methoxy-4a-methyl-1, 2, 3, 4, 4a, 9hexahydrophenanthrene (IV), b.p. 120-125° (air-bath)/ 0.004 mm as a glass (yield 72%). (Calculated for $C_{18}H_{22}O_2$: C, 79.97; H, 8.2; found: C, 79.9; H, 8.2), the 2,4-dinitrophenylhydrazone m.p. 238-240° (dec.) crystallized from acetic acid and contained one molecule of solvent (calculated for $\rm C_{24}H_{26}O_5N_4$, $\rm CH_3COOH\colon N,\,10\cdot97$; found N, 11·13). Mixed melting point with the 2, 4-dinitrophenylhydrazone of the α , β -unsaturated ketone (III) shows 15° depression. Compound IV on hydrogenation with palladium-charcoal (10%) in acetic acid afforded 1, 1-dimethyl-2-keto-6-methoxy-4a-methyl-1, 2, 3, 4, 4a, 9, 10, 10a-octahydrophenanthrene (V), b.p. 180 to $184^{\circ}/0.8$ mm (calculated for $C_{18}H_{24}O_2$: C, 79.4; H, 8.9; found: C, 79.8; H, 8.9); 2,4-dinitrophenylhydrazone, m.p. 198–199° (calculated for $C_{24}H_{28}O_5N_4$: N, 12·3; found: N, 12.3). The keto group in compound V was reduced by CLEMMENSEN's method to give in good yield 1,1-dimethyl-4a-methyl-6-methoxy-1, 2, 3, 4, 4a, 9, 10, 10a-octahydrophenanthrene (DL-6-methoxy podocarpane⁶) (VI), b.p. $145-147^{\circ}/0.8$ mm as a colourless mobile oil; n_{D}^{26} 1.5570 (calculated for $C_{18}H_{28}O$: C, 83.68; H, 10.1; found: C, 83·3; H, 10·2).

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Zusammenfassung

Die Synthese von 1, 1-Dimethyl-4a-methyl-6-methoxy-1, 2, 3, 4, 4a, 9, 10, 10a-octahydrophenanthren (DL-6-Methoxypodocarpan) aus 1-Methyl-7-methoxytetralon-2 wird beschrieben.

- ⁵ G. Cooley, B. Ellis, and V. Petrow, J. chem. Soc. 1955, 2998.
 ⁶ Cf. W. P. Campbell and D. Todd, J. Amer. chem. Soc. 64, 928 (1942).
- ⁷ Present address: Division of Organic Chemistry, National Chemical Laboratory of India, Poona-8, India.

Concentrations of Organic Acids in Animal Tissues *

Insofar as we know there is no summary in the literature of the concentrations of organic acids in animal tissues. Such data are, therefore, presented below.

The references in Table I guide the reader to the appropriate footnote which presents the meaning of the value given. In Table II are shown the sources of the

* Supported in part by a research grant from the National Cancer Institute, National Institutes of Health, U. S. Public Health Service.

¹ G. STORK and W. J. SCHWLENBERG, J. Amer. chem. Soc. 78, 250 (1956).

² P. C. MITTER and S. DE, J. Ind. chem. Soc. 16, 35 (1939).

³ Cf. F. W. Newhall, A. S. Harris, H. W. Frederick, L. E. Johnston, W. J. Richter, E. Walton, N. A. Wilson, and K. Folkers, J. Amer. chem. Soc. 77, 5646 (1955).

⁴ J. W. Cornforth and R. Robinson, J. chem. Soc. 1949, 1855.